

¹³C NMR CHEMICAL SHIFT ASSIGNMENTS OF ALKYL CHAINS OF HISTAMINE ANALOGOUS ω -AMINOALKYLHETEROCYCLES AND BICYCLIC DERIVATIVES

Martin POHL, Wolf-Dieter BLOEDORN, Clemens MUGGE
and Jurgen LIEBSCHER*

Fachbereich Chemie,
Humboldt-Universität Berlin, Hessische Str. 1-2, D-10115 Berlin, Germany

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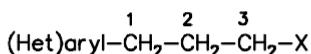
ω -Aminoalkylheterocycles and aromatics as well as bicyclic azolodiazepine derivatives were investigated by means of CH-COSY and INADEQUATE experiments. The ¹³C NMR shifts of all carbon atoms of the aminopropyl and aminobutyl moieties could be assigned. The benzylic position of the ω -aminopropyl moieties appears most upfield rather than more downfield like erroneously reported in the literature.

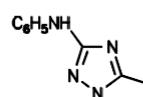
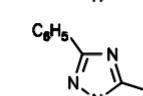
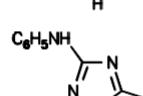
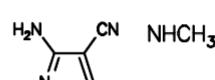
ω -Aminoalkylheterocycles with alkyl chains longer than two carbon atoms such as compounds *I* are homologues and analogues of the naturally occurring histamine. They can be prepared by traditional ring closure reactions¹. Recently a variety of new compounds *I* could be synthesized by ring chain transformation reactions²⁻¹⁰. By derivatization of the ω -amino group of homohistamine *I* (Het)aryl = 4-imidazolyl, X = NH₂) to quanidine moieties highly active H₂-receptor antagonists, such as Impromidine¹¹, were obtained¹²⁻¹⁵. This transformation can be conducted via bicyclic derivatives *II* (refs¹²⁻¹⁵), e.g. *IIa*. ¹³C NMR data of ω -aminoalkylheterocycles *I* were repeatedly reported²⁻⁹, but without a full assignment of the signals of the alkyl side chain (for a ¹³C signal assignment in a branched aminopropylisoxazole see ref.¹⁶. By one exception *IIa* (ref.¹²), the same holds true for bicyclic derivatives *II* (ref.¹⁷). In the former report¹² the central CH₂ group (position 2) of *IIa* was assigned to the most upfield ¹³C signal.

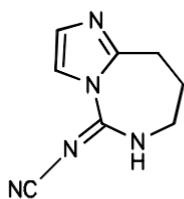
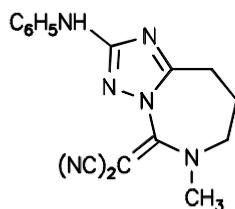
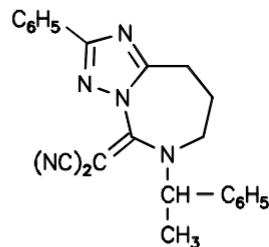
We report now on detailed 2D-NMR experiments which reveal that the previous assignments¹² were erroneous and allow to assign the ¹³C NMR shifts of all carbon atoms of ω -aminoalkylheterocycles *I* and bicyclic derivatives *II*. A number of ω -functionalized alkylbenzenes and heterocycles *I* as well of bicyclic derivatives *II* were in-

* The author to whom correspondence should be addressed.

vestigated by ^1H NMR and by the CH-COSY method (Table I). The assignment of the proton signals was possible based on the signal splitting and chemical shifts. The CH_2 group at position 2 appears most upfield as a multiplet. The two other methylene groups are pseudo triplets with the CH_2 group next to the electronegative heteroatom (position 3) found most downfield. As far as the substituent on nitrogen in vicinity of position 3 in bicyclic derivatives *IIc* is chiral ($-\text{CH}(\text{CH}_3)\text{C}_6\text{H}_5$) the spectra are more complex because of the diastereotopic CH_2 protons. In the aminobutyl compound *Ih* the CH_2 group at position 3 shows a multiplet because of the neighbourhood of an additional methylene group.

*I*

| | (Het)aryl | X | | (Het)aryl | X |
|----------|---|-----------------|----------|--|---|
| <i>a</i> | C_6H_5 | NH_2 | <i>f</i> |  | NHCH_3 |
| <i>b</i> | C_6H_5 | OH | <i>g</i> |  | $\text{NH}-\text{CH}-\text{C}_6\text{H}_5$ CH_3 |
| <i>c</i> | C_6H_5 | Cl | <i>h</i> |  | CH_2NHCH_3 |
| <i>d</i> | C_6H_5 | H | | | |
| <i>e</i> |  | NHCH_3 | | | |

*IIa**IIb**IIc*

Based on the proton chemical shifts, CH-COSY experiments were used to assign the ^{13}C signals. An unambiguous correlation between the highest field shifted inner CH_2 protons (position 2) and the medium shifted ^{13}C signal was found. This proves the location of the carbon signal of position 2 of compounds *I* (*Ih* excepted) and *II* at

TABLE I
Proton and ^{13}C NMR chemical shifts (in ppm, δ -scale) compounds *Ia* – *Ih* and *IIa* – *IIc*. Solvent CD_3SOCD_3

| Compound | ^1H NMR | | | ^{13}C NMR | | |
|---------------------------------|--------------------|--------------------|--------------------|---------------------|--------------------|--------------------|
| | CH ₂ -1 | CH ₂ -2 | CH ₂ -3 | CH ₂ -1 | CH ₂ -2 | CH ₂ -3 |
| <i>Ia</i> | 2.53 ^a | 1.63 ^a | 2.57 ^a | 32.6 | 34.8 | 41.1 |
| <i>Ib</i> | 2.65 | 1.78 | 3.49 | 31.8 | 34.5 | 60.3 |
| <i>Ic</i> | 2.71 | 2.01 | 3.59 | 32.2 | 33.7 | 44.5 |
| <i>Id</i> (ref. ¹⁸) | 2.52 | 1.57 | 0.87 | 37.3 | 24.1 | 13.5 |
| <i>Ie</i> | 2.52 | 1.71 | 2.47 | 24.3 | 27.8 | 50.7 |
| <i>If</i> | 2.65 | 1.78 | 2.49 | 24.0 | 27.5 | 50.9 |
| <i> Ig</i> | 2.87 ^a | 1.90 ^a | 2.57 ^a | 25.5 | 27.2 | 46.8 |
| <i>Ih</i> ^b | 2.63 | 1.70 | 1.45 | 26.0 | 25.3 | 28.7 |
| <i>IIa</i> | 2.92 | 2.01 | 3.20 | 23.9 | 28.3 | 42.1 |
| <i>IIb</i> | 3.05 | 2.30 | 3.40 | 22.1 | 27.2 | 51.2 |
| <i>IIc</i> | 3.06 | 1.48/1.92 | 3.17 | 22.3 | 27.3 | 41.9 |

^a Measured in CDCl_3 . ^b Other signals of $\text{CH}_2\text{--N}$: ^1H 2.46 and ^{13}C 51.2.

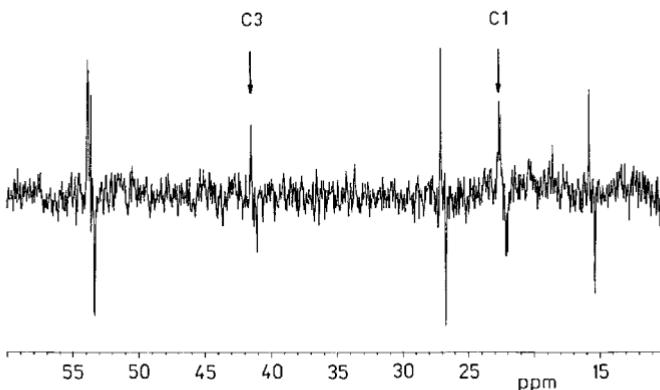


FIG. 1
INADEQUATE spectrum of *IIc*

medium shift of the three propylene ^{13}C signals rather than at the most upfield position, like erroneously reported before¹². An unambiguous assignment of the most upfield signal was possible by an 1D-INADEQUATE NMR spectrum of the bicyclic compound *IIC* (see Fig. 1), which shows the connectivity of CH_2 at position 1 with the ring carbon atom of the triazole ring (two $^1\text{J}(\text{C}, \text{C})$ splittings). Consequently the most downfield signal belongs to the CH_2 group in position 3 (or 4 in *Ih*) next to the heteroatom. The location of the benzylic CH_2 signal (position 1) in compounds *I* and *II* is caused by both, a typical γ -effect of the heteroatom X and the influence of the heterocycle. In all compounds *I* and *II* the same sequence $1 < 2 < 3$ of ^{13}C NMR chemical shifts is found (*Id* X = H excepted), but the shift differences between position 1 and 2 sometimes becomes relatively small (e.g. *Ic*). An analogous sequence of the ^{13}C NMR shifts is found in the aminobutyl compound *Ih* but the additional CH_2 group at position 2 gives rise to the sequence $(2 < 1 < 3 < 4)$.

Most presumable the assignment of ^{13}C NMR shifts of ω -functionalized propyl and butyl substituents can also be applied to other compounds *I* and *II* reported in the literature^{2–10, 12–15}.

EXPERIMENTAL

Compounds *Ia* – *Ic* were purchased from Aldrich. The following compounds were prepared according to literature procedures: *Ie* (ref.⁵), *If* – *Ih* (ref.²), *IIb* and *IIc* (ref.¹⁷). Compound *IIa* was supplied by Dr A. Buschauer, Freie Universität Berlin. All spectra were recorded at ambient temperature on a Bruker AM 300 spectrometer operating at 300.13 and 75.46 MHz for ^1H and ^{13}C , respectively. As solvent CD_3SOCD_3 was used with sample concentrations of 0.1 mol l⁻¹. The ^{13}C – ^1H heteronuclear correlation experiment was carried out with proton decoupling in F_1 CHCORRD (ref.¹⁹) and standard parameters. The INADEQUATE experiments²⁰ were performed with an one bond coupling optimization ($^1\text{J}(\text{C}, \text{C}) = 50$ Hz) without refocussing. The following parameters were used: evolution time for $\text{J}(\text{C}, \text{C})$: 0.01 ($2 \times \text{D}2$) s, repetition time: 10 s, digital resolution: 0.1 Hz/pt.

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